



<b>Study program:</b> Integrated Academic Studies in Pharmacy			
<b>Course title:</b> Quantum Chemistry			
<b>Teacher:</b> Mihalj M. Poša, Zita J. Farkaš-Agatić, Kosta J. Popović			
<b>Course status:</b> elective			
<b>ECTS Credits:</b> 3			
<b>Condition:</b> Organic chemistry I; Organic chemistry II			
<b>Course aim</b> Teach students about the theory of quantum chemistry that is used for calculating the density distribution of electrons in the molecule, and the parameters resulting from the distribution of electrons in order to be used as molecular descriptors			
<b>Expected outcome of the course:</b> Quantum nature of the distribution of electrons in multinuclear systems. Students will be able to independently using appropriate software to calculate molecular descriptors derived from the distribution of electrons.			
<b>Course description</b> <i>Theoretical education</i> 1. Wave function 2. Born Openchaimer approximation 3. Valence connection theory 4. Molecule orbit theory 5. Walsh diagram 6. Huckel method 7. Semi empirical methods 8. Ab initio methods 9. Application: Solvation Energy  <i>Practical education</i> Usage of proper software			
<b>Literature</b> <i>Compulsory</i> 1. Grant GH, Richards WG. Computational Chemistry, Oxford University Press, 1955			
<b>Number of active classes</b>	<b>Theoretical classes:</b> 30	<b>Practical classes:</b> 15	
<b>Teaching methods</b> Lectures, practice			
<b>Student activity assessment</b> (maximally 100 points)			
<b>Pre-exam activities</b>	<b>points</b>	<b>Final exam</b>	<b>points</b>
Lectures		Written	
Practices		Oral	40
Colloquium		.....	
Essay	60		